

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

Reactions With Hydrazonoyl Halides 59¹: Synthesis and Antimicrobial Activity of 2,3-Dihydro-1,3,4-thiadiazole, Triazolino[4,3-*a*]pyrimidine, and Pyrimido[1,2-*b*][1,2,4,5]tetrazin-6-one Containing Benzofuran Moiety

Abdou O. Abdelhamid^a; Mahmoud A. Mohamed^b; Yasser H. Zaki^c

^a Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt ^b Department of Textile, Faculty of Industrial Education, Beni-Suef University, Egypt ^c Department of Chemistry, Faculty of Science, Beni-Suef University, Egypt

To cite this Article Abdelhamid, Abdou O. , Mohamed, Mahmoud A. and Zaki, Yasser H.(2008) 'Reactions With Hydrazonoyl Halides 59¹: Synthesis and Antimicrobial Activity of 2,3-Dihydro-1,3,4-thiadiazole, Triazolino[4,3-*a*]pyrimidine, and Pyrimido[1,2-*b*][1,2,4,5]tetrazin-6-one Containing Benzofuran Moiety', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 183: 7, 1746 — 1754

To link to this Article: DOI: 10.1080/10426500701734265

URL: <http://dx.doi.org/10.1080/10426500701734265>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Reactions With Hydrazonoyl Halides 59¹: Synthesis and Antimicrobial Activity of 2,3-Dihydro-1,3,4-thiadiazole, Triazolino[4,3-*a*]pyrimidine, and Pyrimido[1,2-*b*][1,2,4,5]tetrazin-6-one Containing Benzofuran Moiety

Abdou O. Abdelhamid,¹ Mahmoud A. Mohamed,²
and Yasser H. Zaki³

¹Department of Chemistry, Faculty of Science, Cairo University,
Giza, Egypt

²Department of Textile, Faculty of Industrial Education, Beni-Suef
University, Egypt

³Department of Chemistry, Faculty of Science, Beni-Suef University,
Egypt

*2,3-Dihydro-1,3,4-thiadiazole, triazolino[4,3-*a*]pyrimidine and pyrimido[1,2-*b*][1,2,4,5]tetrazin-6-one containing benzofuran Moiety were synthesized from C-benzofuran-2-yl-N-phenylhydrazonoyl bromides, and the appropriate alkyl arylidenehydrazinacabodithioates and pyrimidine-2-thione and N-aminopyrimidine-2-thione, respectively. All structures of the newly synthesized compounds were elucidated by elemental analysis, spectral data, and alternative synthetic methods whenever possible. Newly compounds are capable of high inhibiting the growth of bacteria (gram positive and gram negative).*

Keywords 2,3-Dihydro-1,3,4-thiadiazole; hydrazonoyl bromide; pyrimido[1,2-*b*][1,2,4,5]tetrazin-6-one; triazolino[4,3-*a*]pyrimidine

INTRODUCTION

Benzofuran are very important compounds due to their broad spectrum of biological and pharmacological effects. Benzofuran are considered non-steroidal anti-inflammatory drugs (NSAID), where the action of (NSAID) is lowering the prostaglandin production through inhibition of cyclooxygenase (COX). Benzofuran are among the COX-2 inhibitors.^{2,3} In addition, diverse pharmacological properties have been associated with benzofuran derivatives.^{4–8} These include pesticidal,⁹ fungicidal,

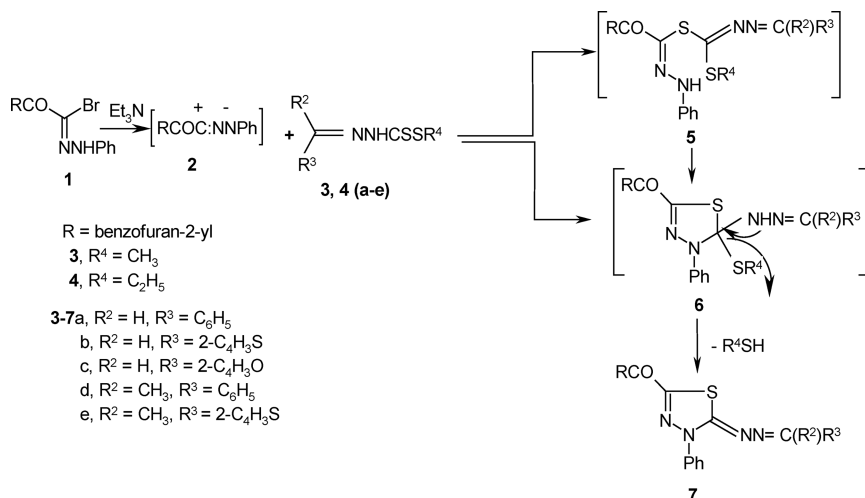
Received 17 July 2007; accepted 4 September 2007.

Address correspondence to Abdou O. Abdelhamid, Department of Chemistry, Faculty of Science, Cairo University, Giza 12613, Egypt. E-mail: abdelhamid45@gmail.com

antimicrobial, antioxidant,¹⁰ anti-inflammatory,¹¹ antihistaminic,¹² antiallergic,¹³ antitumor,¹⁴ anticonvulsant, and antinociceptive¹⁵ agent. We report here several heterocyclic compounds benzofuran moiety expected to possess biological activity.

RESULTS AND DISCUSSION

Treatment of *C*-(2-benzofuranyl)-*N*-phenylhydrazonoyl bromide (**1**) with the appropriate alkyl carbodithioates^{16–18} **3(a–e)** or **4(a–e)** in ethanol containing triethylamine afforded 2,3-dihydro-1,3,4-thiadiazoles **7(a–e)**, respectively (Scheme 1). Structures **7** were confirmed by elemental analysis, spectral data (*cf.* Experimental).

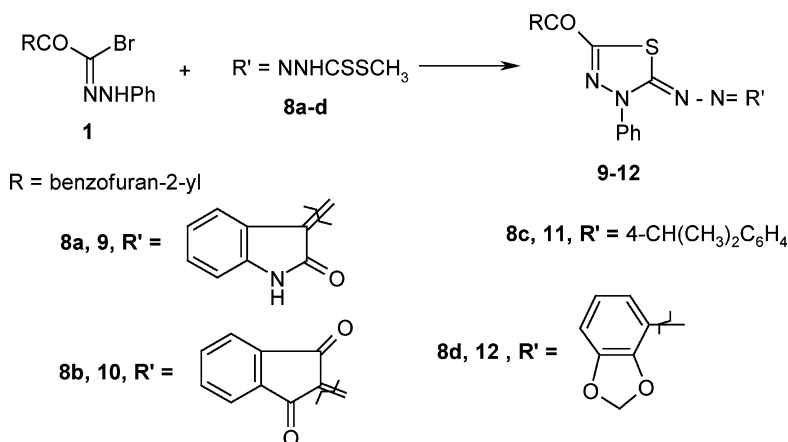


SCHEME 1

In the light of the foregoing results, the mechanism outlined in Scheme 1 seems to be the most plausible pathway for the formation of **7** from the reaction of the **1** with **3** or **4**. The reaction involves initial formation of thiohydrazone **5**, which undergoes to yield the intermediate **6** or via 1,3-dipolar cycloaddition of nitrilimine **2**, (which was prepared in situ from **1** with triethylamine) to C=S double bond of **3** (or **4**). The formation of **5** and **6** are similar to the reaction of hydrazonoyl chloride with 1-phenyl-1,4-dihydrotetrazole-5-thione¹⁹ and 5-phenyl-1,3,4-thiadiazole-2(3H)-thione.²⁰

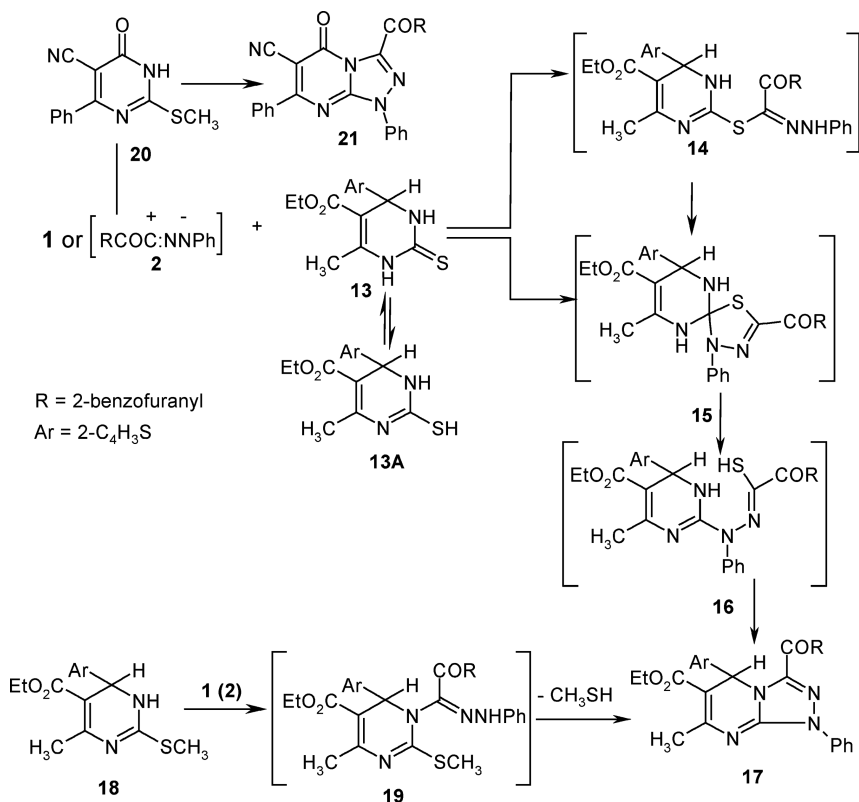
Analogously, **1** reacted with each of 3-{aza-[(methylthioxomethyl)-aminomethylene]indoline-2-one,²¹ 3-{aza-[(methylthioxomethylamino]

methylene}-indan-1,3-dione,²² methyl 1-(4-isopropylphenyl)ethylidenedithiohydrazono-carbodithioate²³ and methyl, (1,3-benzodioxol-4-yl)ethylidenedithiohydrazono-carbodithioate²³ to give 2-{1,2-diaza-2-[5-(benzo[d]furan-2-yl)carbonyl]-3-phenyl-(1,3,4-thiadiazolin-2-ylidene)}indane-1,3-dione (**9**), 3-{1,2-diaza-2-[5-(benzo[d]furan-2-yl)carbonyl]-3-phenyl-(1,3,4-thiadiazolin-2-ylidene)}indolin-2-one (**10**), 2-{1,2-diaza-3-[4-(methylethyl)phenyl]prop-2-enylidene}-3-phenyl(1,3,4-thiadiazolin-5-yl)benzo[d]furan-2-ylketone (**11**) and 2-(3-(2H)-benzo[d]1,3-dioxolan-4-yl)-1,2-diazaprop-2-enylidene)-3-phenyl(1,3,4-thiadiazolin-5-yl)benzo[d]furan-2-ylketone (**12**) (Scheme 2).



SCHEME 2

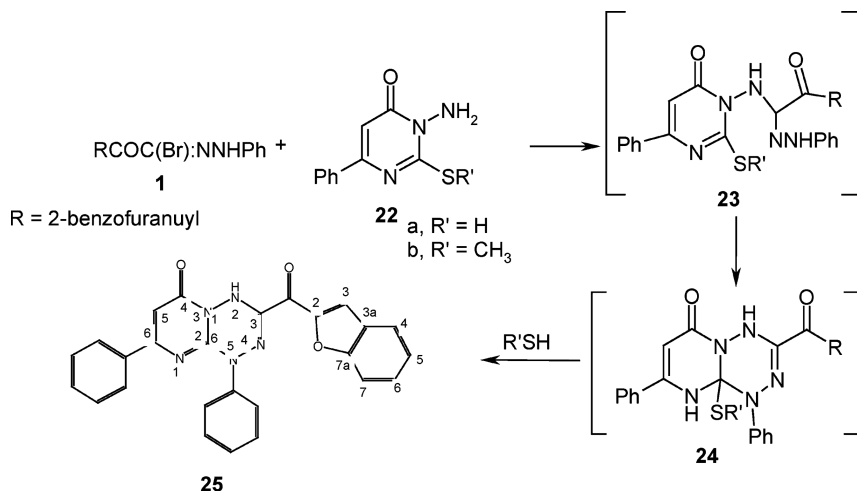
Also, treatment of **1** with each of the pyrimidine-2-thione **13** and **20** in boiling chloroform to give triazolino[4,3-*a*]pyrimidines in a good yields **17** and **21**, respectively (Scheme 3). The structure of **17** was elucidated by elemental analysis, spectral data, and alternative synthesis route. Thus, ¹H NMR spectrum of **17** showed signals at δ = 1.23 (t, 3H, CH₂CH₃), 2.56 (s, 3H, CH₃), 4.09 (q, 2H, CH₂CH₃), 5.05 (s, 1H, pyrimidine H-4), 7.16–7.25 (m, 3H, thiophene protons), 7.44–7.72 (m, 7H, ArH's), 8.05 (s, 1H, benzofuran H-4), and 8.24 (d, *J* = 8Hz, 2H, ArH's). Its IR spectrum revealed bands at 1702 (CO ester), 1650 (CO conjugated) and 1615 (C=N). Also, compound **17** was obtained from the reaction of ethyl 6-methy-2-methylthio-4-(2-thienyl)-3,4-dihydropyrimidine-5-carboxylate **18** with **1** in boiling sodium ethoxide solution. The mechanism outlined in Scheme 3 seems to be the most plausible pathway for the formation of **17** from the reaction of **1** with **13** or **18**. Two possible pathways can account for the formation **17**: 1)- 1,3- addition of the thiol tautomer **13A** to the nitrilium imide **2** to give the thiohydrazonate ester **14** which undergoes



SCHEME 3

nucleophilic cyclization to yield spiro compounds **15**. The latter ring open to **16** which cyclized to yield **17** by loss hydrogen sulfide; and 2)-1,3-cycloaddition of nitrilium imide **2** to C=S double bond of **13** can give directly **16** (Scheme 3). All attempts to isolate any intermediates are unsuccessful.

Treatment of **1** with 3-amino-6-phenyl-2-thioxo-2,3-dihydro-1*H*-pyrimidine-4-one²⁴ (**22**) in boiling ethanol containing triethylamine gave 3-(4-methyl-2-phenyl)thiazol-5-yl-1,7-diphenyl-4*a*-hydro-4*H*-pyrimidino[1,2-*b*]1,2,4,5-tetrazin-5-one (**25**), respectively (Scheme 4). Structure **25** was based on spectral and microanalysis data. IR spectrum of **25** revealed bands at 3290 and 1680 cm^{-1} due to NH and CO groups, respectively. Its ^1H NMR spectrum showed signals at $\delta = 6.84$ (s, 1H, pyrimidine H-5), 6.42-7.54 (m, 15H, ArH's) and 8.52 (s, br., 1H, NH). ^{13}C NMR spectrum of **25** showed 23 signals: $\delta = 152$ (benzofuran



SCHEME 4

C-2), 116 (benzofuran C-3), 131 (benzofuran C-3a), 121, 123, 124, 111 (benzofuran C-4, C-5, C-6, C-7), 161 (benzofuran C-7a), 154 (tetrazine C-3), 163 (tetrazine C-5 or pyrimidine C-2), 110 (pyrimidine C-5), 156 (pyrimidine C-6), 126, 128, 136 (phenyl attach pyrimidine ring), 116, 118, 129, 144 (phenyl attach to tetrazine ring), 159, 178 (2 C=O).

Formation of **25** from reaction of hydrazonoyl bromide **1** with either **22a** or **22b**, it is suggested that the reaction starts with the formation of amidrazone **23** followed by cyclization to **24** to give the product **25** via elimination of hydrogen sulfide or methanethiol (Scheme 4).

Antimicrobial Activity

The tested microorganism was gram +ve bacteria, gram -ve bacteria and some Fungal-plant. Sensitivity of the selected microorganisms to some synthesized compounds were determined in vitro culture that were dissolved in chloroform, the tests were carried out using the filter paper and hole plate method.^{25,26} Studies on the biological activity of compounds in comparison with Chlorumphenicol and Terbinafin showed in Table I. In general, all tested compounds were capable of a high inhibiting the growth of gram positive and gram negative. Also, showed the tested compounds were a high inhibition towards *Candida albicans* (Fungus) and negative *Aspergillus flvus* (Fungus).

TABLE I Response of Various Microorganisms to Some Synthesized Compounds in vitro (Culture)

Microorganisms /compound no.	<i>Bacillus</i>				<i>Aspergills Candida</i>	
	<i>subtilis</i> (G ⁺)	<i>Echerichia coli</i> (G ⁻)	<i>Staphylococcus albus</i> (G ⁺)	<i>Streptococcus faecalis</i> (G ⁺)	<i>flvus</i> (Fungus)	<i>albicans</i> (Fungus)
7a	13	14	14	13	0.00	14
7b	13	13	14	12	0.00	15
7c	13	13	14	13	0.00	13
7d	12	13	11	13	0.00	12
7e	13	12	13	13	0.00	13
9	12	13	12	13	0.00	13
10	12	13	14	13	0.00	13
11	12	13	13	14	0.00	12
12	13	14	14	13	0.00	14
17	12	12	13	13	0.00	13
21	12	12	13	13	0.00	13
25	12	12	12	12	0.00	12

Reference standard; chlorumphinecol was used as a standard antibacterial agent; and terbinafin was used as a slandered antifungal agent. Values show zone of inhibition in mm. Diameter of the inhibition zones were: high (11–15 mm), moderate (6–10 mm), slight (1–5 mm), and negative (0).

EXPERIMENTAL

All melting points were uncorrected. IR spectra were recorded (KBr disc) on a Shimadzu FT-IR 8201 PC Spectrophotometer. ¹H-NMR spectra were recorded in CDCl₃ or (CD₃)₂SO on a Varian Gemini 200 MHz Spectrometer and chemical shifts were expressed in units using TMS as an internal reference. Elemental analyses were carried out at the Microanalytical Center, Cairo University, Giza, Egypt, and National Research Centre. Hydrazonoyl bromide **1** was prepared as previously reported in literature.²⁷

2,3-Dihydro-1,3,4-Thiadiazoles **7a–e** and **9–12**

Triethylamine (0.5 g (0.75 ml), 5 mmol) was added dropwise with stirring to a mixture of the appropriate alkyl carbodithioates **3a–e** (or **4a–e**), **8a–d** (5 mmol) and **1** (1.8 g, 5 mmol) in ethanol (20 mL). The resulting solid, which formed after 30 min, was collected and recrystallized from dioxin-ethanol to give 2,3-dihydro-1,3,4-thiadiazoles **7a–e** and **9–12**, respectively, in a good yield (Tables II and III).

TABLE II Characterization Data of the Newly Synthesized Compounds

Comp. no.	Mp. °C (solvent)	Color yield (%)	Mol. Formula (mol. wt.)	Calcd./found (%)			
				C	H	N	S
7a	210	Red	C ₂₄ H ₁₆ N ₄ O ₂ S	67.91	3.80	13.20	7.55
	Dioxan-EtOH	87	424.48	67.71	3.90	13.02	7.42
7b	195	Orange	C ₂₂ H ₁₄ N ₄ O ₂ S ₂	61.38	3.28	13.01	14.90
	Dioxan-EtOH	85	430.51	61.45	3.00	13.24	15.10
7c	190	Red	C ₂₂ H ₁₄ N ₄ O ₃ S	63.76	3.40	13.52	7.74
	Dioxan-EtOH	82	414.45	63.67	3.04	13.32	7.54
7d	190	Orange	C ₂₅ H ₁₈ N ₄ O ₂ S	68.48	4.14	12.78	7.31
	Dioxan-EtOH	78	438.51	68.35	4.10	12.87	7.51
7e	180	Brown	C ₂₃ H ₁₆ N ₄ O ₂ S ₂	62.14	3.63	12.60	14.43
	Dioxan-EtOH	80	444.54	62.10	3.52	12.41	14.34
9	264	Red	C ₂₅ H ₁₅ N ₅ O ₃ S	64.51	3.25	15.04	6.89
	Dioxan-EtOH	84	465.49	64.72	3.35	14.95	6.75
10	>300	Yellow	C ₂₆ H ₁₄ N ₄ O ₄ S	65.27	2.95	11.71	6.70
	Dioxan-EtOH	79	478.49	65.42	2.75	11.92	6.85
11	175	Orange	C ₂₇ H ₂₂ N ₄ O ₂ S	69.51	4.75	12.01	6.87
	Dioxan-EtOH	82	466.57	69.40	4.75	11.95	6.68
12	220	Brown	C ₂₅ H ₁₆ N ₅ O ₄ S	64.09	3.44	11.96	6.84
	Dioxan-EtOH	83	468.49	64.21	3.54	12.10	6.95
17	158	Orange	C ₂₈ H ₂₂ N ₄ O ₄ S	65.87	4.34	10.97	6.28
	EtOH	68	510.58	65.70	4.21	10.72	6.48
21	285–290	Yellow	C ₂₇ H ₁₅ N ₅ O ₃	70.89	3.31	15.31	—
	Dioxan	74	457.45	70.84	3.55	15.35	—
25	214–216	Red	C ₂₆ H ₁₇ N ₅ O ₃	69.79	3.83	15.65	—
	EtOH	71	447.46	69.54	3.65	15.87	—

1,2,4-Triazolo[4,3-*a*]pyrimidines **17**, **21**, and Pyrimidino[1,2-*b*]1,2,4,5-tetrazin-5-one **25**

Method A

An equimolar amount of the hydrazoneyl bromide **1**, the appropriate pyrimidine-2-thione **13** (or **20**), **22b** and sodium ethoxide (5 mmol) in ethanol (20 mL) was refluxed for 3 h. The reaction mixture was cooled and the resulting solid was collected and recrystallized from ethanol to give **17**, **21**, and **25**, respectively (Tables II and III).

Method B

A mixture of the hydrazoneyl bromide **1** (1.8 g, 5 mmol), **13** (1.48 g, 5 mmol) or **22a** (1.08 g, 5 mmol) and triethylamine (0.5 g, (0.75 mL), 5 mmol) in chloroform (20 mL) containing was refluxed for 10 hrs. Chloroform was evaporated under reduced pressure and the residue solid

TABLE III Spectral Data of Some Newly Synthesized Compounds

Comp. no	Spectral data
7a	$^1\text{H NMR}$: δ = 7.01–7.59 (m, 15 H, ArH's) and 8.47 (s, 1H, CH=).
7b	$^1\text{H NMR}$: δ = 6.48–7.59 (m, 13 H, ArH's) and 8.35 (s, 1H, CH=).
7c	$^1\text{H NMR}$: δ = 6.48–7.59 (m, 13 H, ArH's) and 8.42(s, 1H, CH=).
7d	$^1\text{H NMR}$: δ = 2.12 (s, 3H, CH ₃), 6.45–7.85 (m 15 H, ArH's).
7e	$^1\text{H NMR}$: δ = 2.12 (s, 3H, CH ₃), 6.45–7.85 (m 13 H, ArH's).
9	$^1\text{H NMR}$: δ = 6.46–7.82 (m, 19H, ArH's), 9.23 (s, br., 1H, NH).
10	$^1\text{H NMR}$: δ = 7.39–8.37 (m, ArH's).
11,	$^1\text{H NMR}$: δ = 1.26 (d, 6H, J = 8 Hz, (CH ₃) ₂ CH, 2.95 (sept., 1H, (CH ₃) ₂ CH), 7.31–7.81 (m, 10 H, ArH's), 8.05 (d, 2H, ArH's), 8.24 (s, 1H, benzofuran H-4), 8.42 (s, 1H, CH=).
12	$^1\text{H NMR}$: δ = 5.02 (s, 2H, OCH ₂ O), 6.85 (d, 1H), 7.15 (d, 1H), 7.26–7.64 (m, 9 H), 7.77 (d, 1H), 8.04 (d, 1H), 8.23 (s, 1H, CH=).
17	$^1\text{H NMR}$: δ = 1.23 (t, 3H, CH ₂ CH ₃), 2.56 (s, 3H, CH ₃), 4.09 (q, 2H, CH ₂ CH ₃), 5.05 (s, 1H, pyrimidine H-4), 7.16–7.25 (m, 3H, thiophene protons), 7.44–7.72 (m, 7H, ArH's), 8.05 (s, 1H, benzofuran H-4) and 8.24 (d, J = 8Hz, 2H, ArH's).
21	$^1\text{H NMR}$: δ = 6.46–7.82 (m, ArH's).
25	$^1\text{H NMR}$: δ = 6.84 (s, 1H, pyrimidine H-5), 6.46–7.72 (m, 15H, ArH's), 8.52 (s, 1H, NH).

was crystallized from ethanol to give products identical in all aspects (m.p., mixed m.p., and spectra) with corresponding products obtained by Method A.

REFERENCES

- [1] Part 58: A.O. Abdelhamid, Z. H. Ismail, and A. Abdel-Aziem., *Phosphorus, Sulfur, and Silicon*, **183**, 1735–1745 (2008).
- [2] F. D. Popp, *J. Heterocycl. Chem.*, **19**, 589 (1982).
- [3] K. M. Basavaraga, Y. S. Agasimundin, K. M. Mahadevan, and V. P. Vaidya, *Indian I. Heterocycl. Chem.*, **13**, 155 (2003).
- [4] K. T. Vinh, M. Ahmadi, L. O. P. Delgado, F. S. Prerez, M. H. Walters, J. H. Smith, J. P. Nichols, and C. Simons, *Bioorg. Med. Chem. Lett.*, **9**, 2105 (1999).
- [5] Z. Tomaszewski, P. M. Johnson, X. Haung, and E. D. Nichols, *J. Med. Chem.*, **35**, 2061 (1992).
- [6] W. J. Ellingboe, R. T. Alessi, M. T. Dolak, T. T. Nguyen, D. J. Tomer, F. Guzzo, and L. M. Mccaleb, *J. Med. Chem.*, **35**, 1176 (1992).
- [7] P. G. Wyatt, M. J. Allen, J. Chilcott, C. J. Grandier, D. G. Livermore; J. E. Mordaunt, F. Nerozzi, M. Patel, M. J. Perren, G. G. Weingarten, S. Shabbir, P. M. Woollard, and P. Zhou, *Bioorg. Med. Chem. Lett.*, **12**, 1405 (2002).
- [8] F. Yoneda, T. Moto, M. Sakae, H. Ohde, B. Knoll, and I. Miklya, *Bioorg. Med. Chem.*, **9**, 1197 (2001).
- [9] F. A. John, B. Sentsetsa, L. Guoqiang, and S. Michelle, *J. Nat. Prod.*, **60**, 1214 (1977).
- [10] R. S. Wards, *Nat. Prod. Rep.*, **16**, 75 (1999).

- [11] L. Santana, M. Teijeira, E. Uriarte, C. Teran, B. Linares, R. Villar, R. Laguna, and E. Cano, *Eur. J. Pharm. Sci.*, **7**, 161 (1999).
- [12] A. Leonardi, G. Nava, and D. Nardi, *Farmaco. Ed. Sci.*, **38**, 290 (1983).
- [13] J. H. Musser, R. E. Brown, B. Love, K. Bailey, H. Jones, R. Kahen, F. C. Huang, A. Khandurala, M. Leibowitz, P. S. Goldman, and D. Donigi-Ruzza, *J. Med. Chem.*, **27**, 121 (1984).
- [14] I. Hayakawa, R. Shioya, T. Agatsuma, H. Furukawa, and Y. Sugano, *Bioorg. Med. Chem. Lett.*, **14**, 3411 (2004).
- [15] K. M. Dawood, H. Abdel-Gawad, E. A. Rageb, M. Ellithey, and H. A. Mohamed, *Bioorg. Med. Chem.*, **14**, 3672 (2006).
- [16] J. Sandstrom, *Arkiv Kemi.*, **9**, 225, 1956; *Chem. Abstr.*, **50**, 15516d (1956).
- [17] J. Sandstrom, *Acta Chem., Scand.*, **17**, 937, 1963; *Chem. Abstr.*, **60**, 10072f (1963).
- [18] J. Korosi, *Ger. Offen.* 1, 934, 809 29 Jan., 1970; *Chem. Abstr.*, **72**, 100334s (1970).
- [19] R. Huisgen, R. Garashey, M. Seidal, H. Knupfer, and R. Schmidt, *Ann. Chem.*, **658**, 169 (1962).
- [20] R. N. Butler, E. P. Ni Bhraidaigh, and K. J. Fitzgerald, *J. Chem. Res.*, (S), **306** (1993); *J. Chem. Res.*, (M), (1948).
- [21] Y. P. Kovtun and N. N. Ramanov, *Khim., Geterosikt, Soedin*, 211 (1985).
- [22] A. O. Abdelhamid, S. M. Abdel-Gwad, and S. F. El-Sharnoby, *Phosphorus, Sulfur, Silicon, and Related Elem.*, **177**, 2699 (2002).
- [23] Y. H. Zaki, S. A. Ahmed, A. M. Hussein, and A. O. Abdelhamid, *Phosphorus, Sulfur, Silicon, and Related Elem.*, **181**, 825 (2006).
- [24] A. S. Shawali, M. A. Abdallah, and M. M. Zayed, *J. Heterocyclic Chem.*, **39**, 45 (2002).
- [25] C. Refer Lefert, H. Siripumchidbouree, S. Hamspons, S. Workman, D. Sigee, H. A. S. Epton, and A. Harbour, *J. Appl. Bact.*, **78**, 97 (1955).
- [26] W. E. Solomons and N. J. Doorenbos, *J. Pharm. Sci.*, **63**, 19, (1974).
- [27] A. O. Abdelhamid, F. A. Attaby, and M. Y. Zaki, *Phosphorus, Sulfur, Silicon, and Related Elem.*, **53**, 403 (1990).